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In Situ Redispersion of Platinum Nanoparticles Supported on Ceria-Based Oxide for Auto-Exhaust Catalysts

Improving The lifetime of auto-exhaust catalysts?



Automotive Catalyst



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OUTLINE

1. Background

Automotive Catalyst

2. Experimental

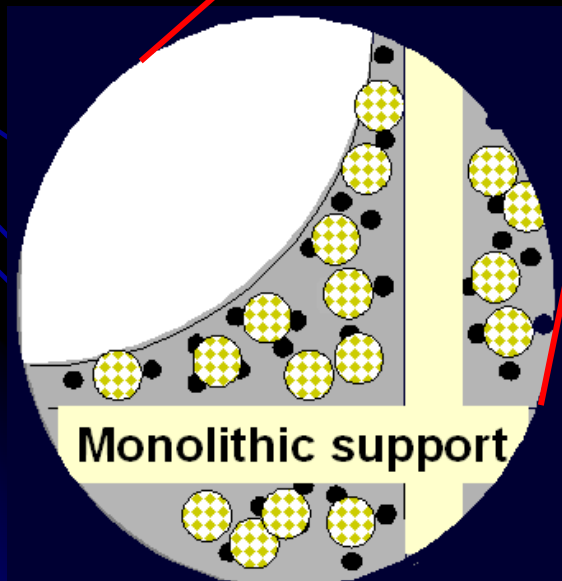
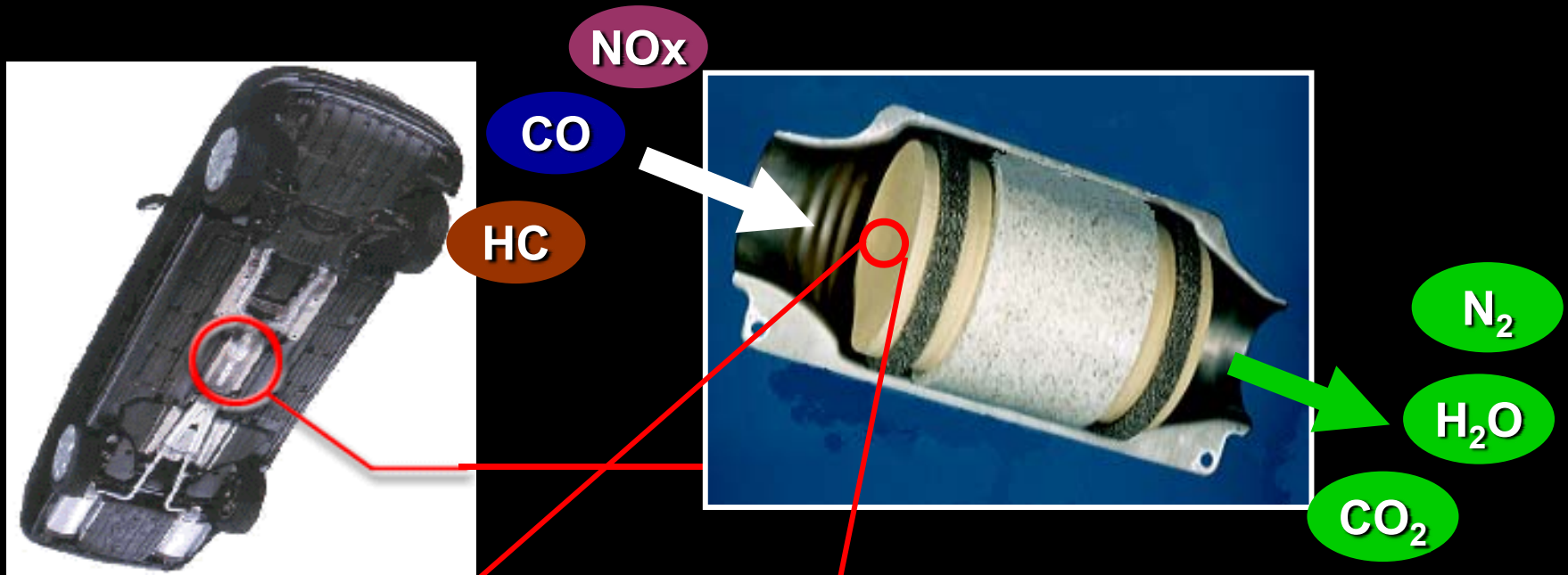
3. Results

- a. in-situ Turbo-XAS in fluorescence mode
- b. in-situ dynamic measurement of Pt particle size
- c. Dynamic observation of Pt sintering/redispersion
- d. in-situ TEM movie of Pt redispersion

4. Summary

5. Future view

Automotive Three-way Catalyst (TWC)



○ Active site : Pt, Rh, Pd

■ Support : alumina, etc

● Additives : CeO₂, etc

Exhaust Condition from Gasoline Engine



**Gasoline
Engine**

Exhaust



Catalytic Converter



Temperature; up to 1000 °C

In-situ dynamic observation of precious metals is important to the development of advanced catalysts.



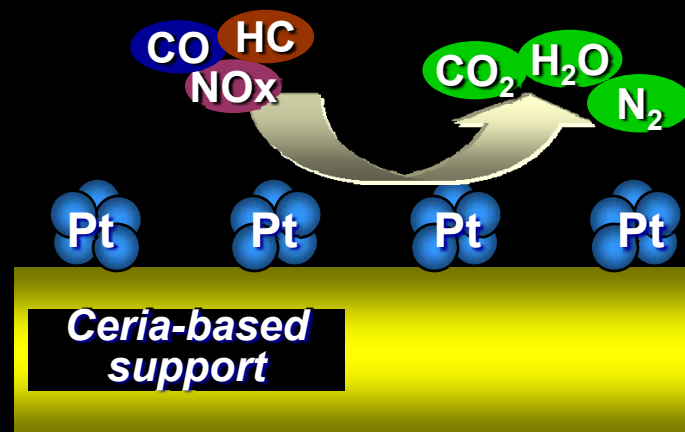
Pt Sintering/Redispersions (Static study)

Reductive

Redispersions



Sintering



Our Goal:

In-situ dynamic observation of the Pt

Sintering/Redispersions in simulated exhaust flowing.



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Experimental

➤ Catalyst

2 wt% Pt/ γ -Al₂O₃

2 wt% Pt/CZY(Ce-Zr-Y mixed oxide)

CZY: 50 wt% CeO₂, 46 wt% ZrO₂ and 4 wt% Y₂O₃

➤ XAFS measurement

Pt L₃-edge XANES every 1~6 sec

ESRF ID24

Turbo-XAS (variant of DXAFS) in fluorescence mode

in-situ: Simulated gas

O₂(4 or 20%)/He ↔ H₂(3%)/He cyclically, at 400~800°C

➤ In situ TEM Observation

Univ. Arizona

820 °C, 10 torr O₂ ,



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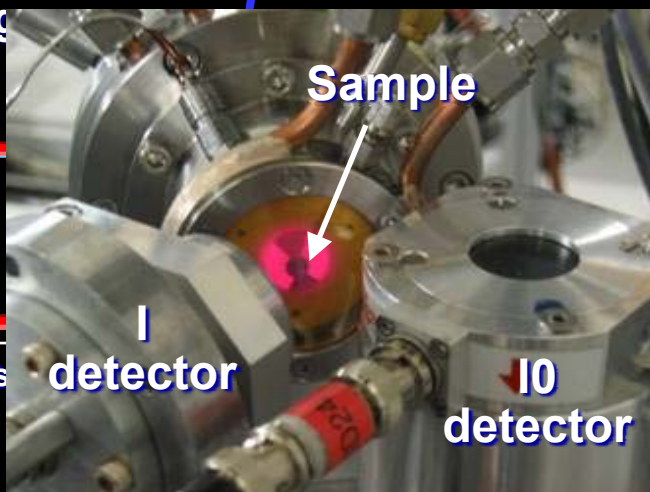
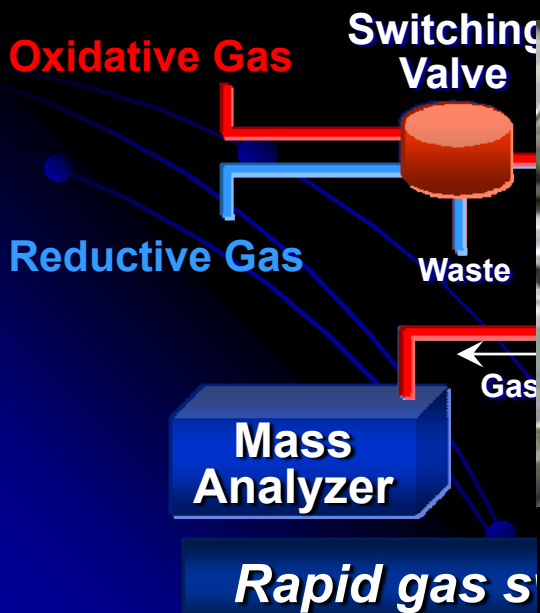
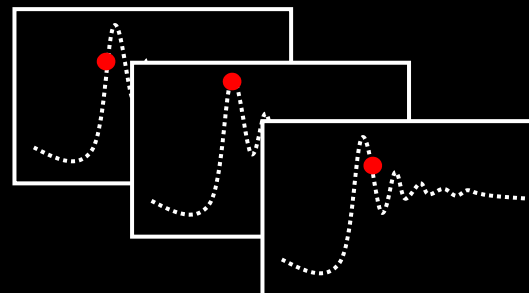
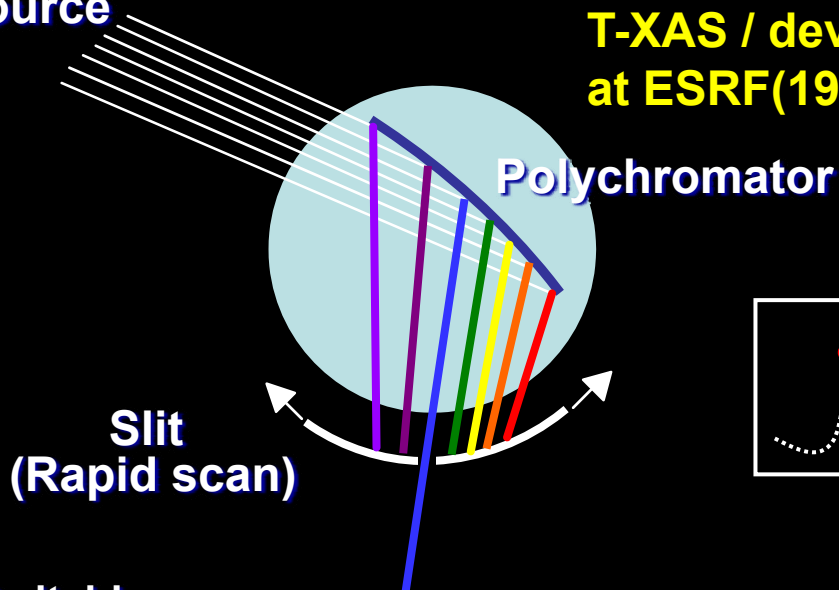
5. Future view



in-situ Turbo-XAS in Fluorescence Mode

T-XAS / developed by S. Pascarelli et al. at ESRF(1998)

Undulator Source



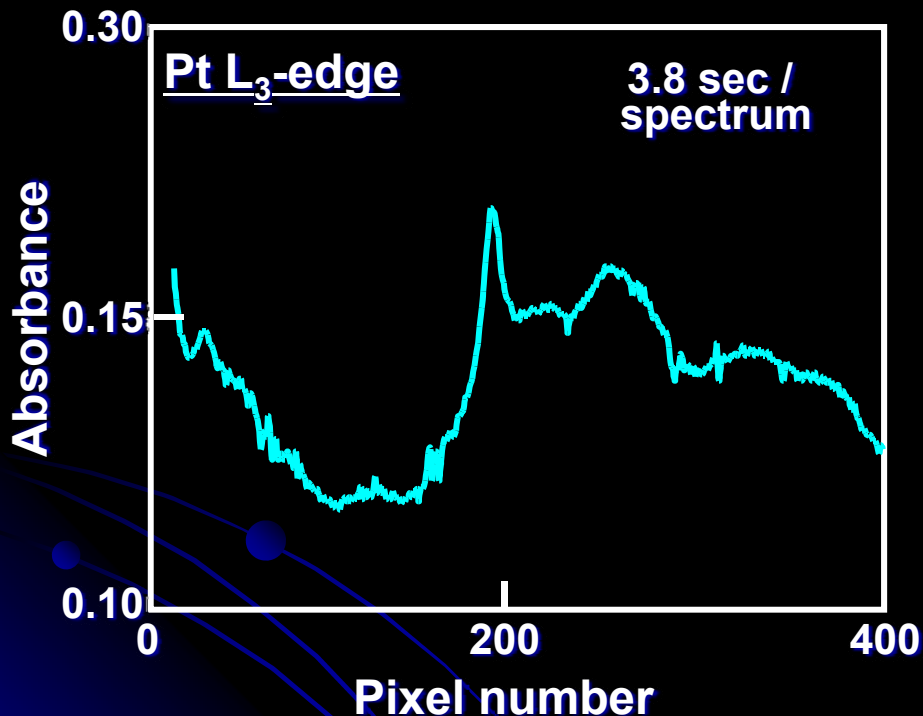
Very fine time resolution for fluorescence XANES

Manufactured by B. Gorges from the ESRF Sample Environ. Lab

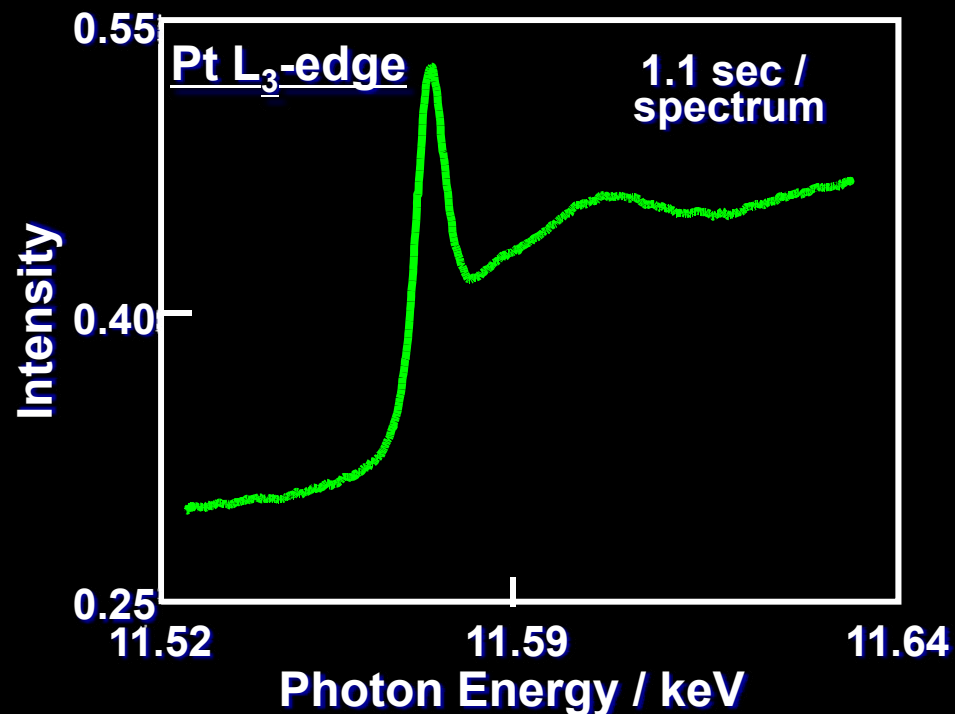
Comparison of Spectra on Pt/CZY

2 wt% Pt/CZY (Ce-Zr-Y mixed oxide)

DXAFS in Transmission mode
at ESRF ID24



T-XAS in fluorescence mode
at ESRF ID24



High loading amount of heavy elements such as Ce and Zr in this sample strongly absorbs the X-ray.

Fluorescence T-XAS permits time-resolved measurements.



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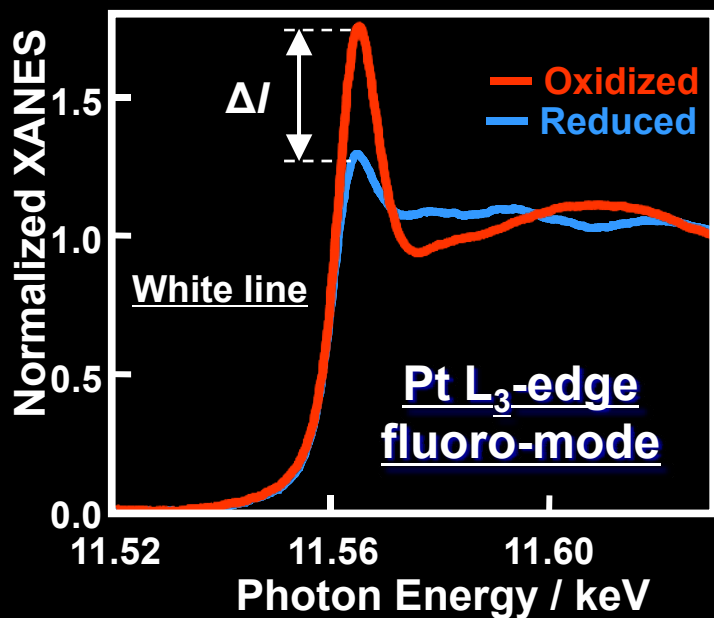
d. in-situ TEM movie of Pt redispersion

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How to Analyze Pt Particle Size?

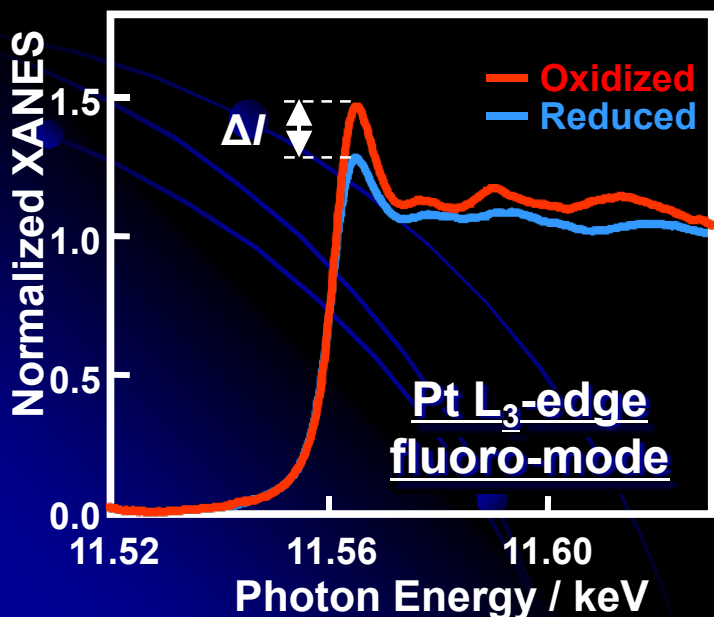
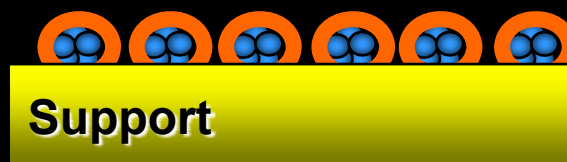


Small Pt particles → Large ΔI

Oxidized

Surface Pt oxide

Pt metal

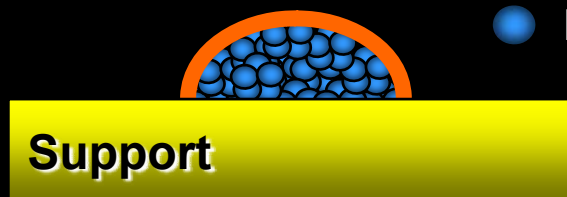


Large Pt particles → Small ΔI

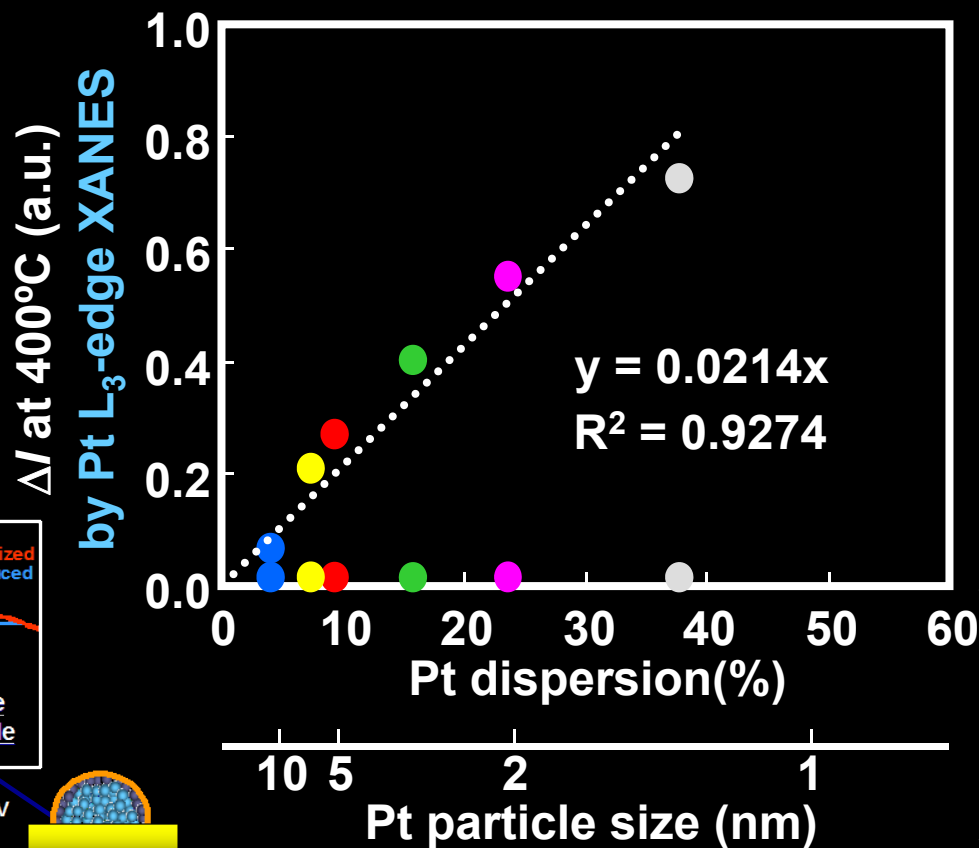
Oxidized

Surface Pt oxide

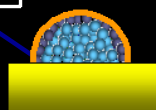
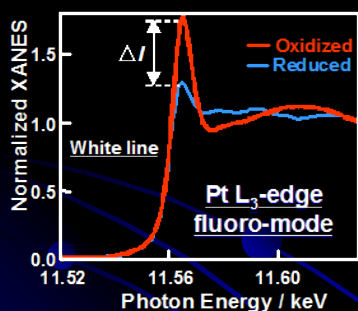
Pt metal



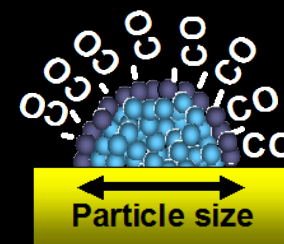
Relationship between ΔI and Pt Particle Size



The Pt/CZY catalyst were preliminary aged at various temperature to obtain specific particle size of sintered Pt.



by CO chemisorption method (ex-situ, off-line)



The good linear relationship between Pt dispersion and ΔI suggests that the Pt particles are oxidized only on the surface region.

By using fluorescence T-XAS with ~ 1 sec time-resolution (ΔI), it is possible to track the change in Pt particle size under in-situ condition.

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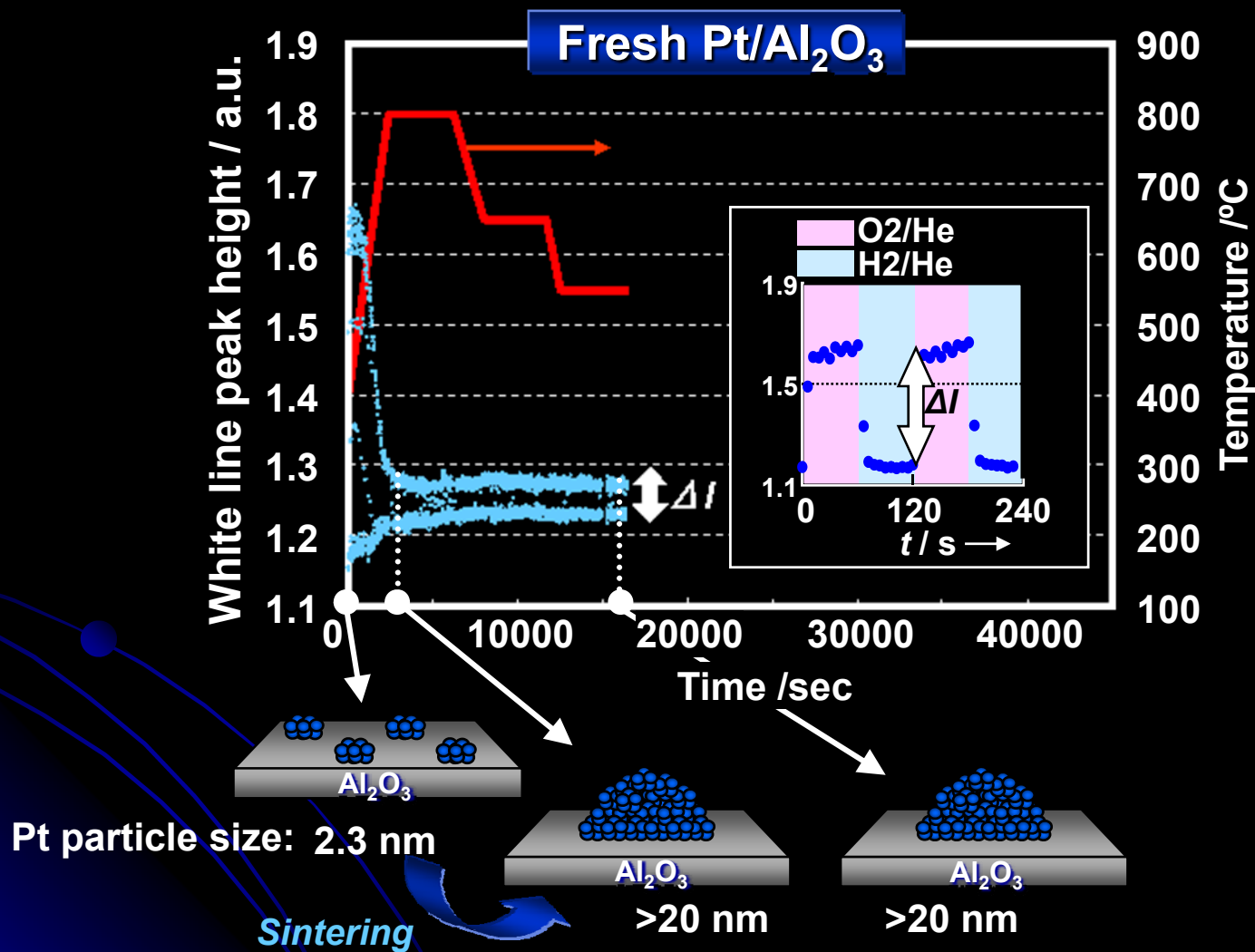
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In-situ Dynamic Observation in Pt/Al₂O₃

3% H₂/He(60sec) ↔ 20 or 4 % O₂/He(60sec)



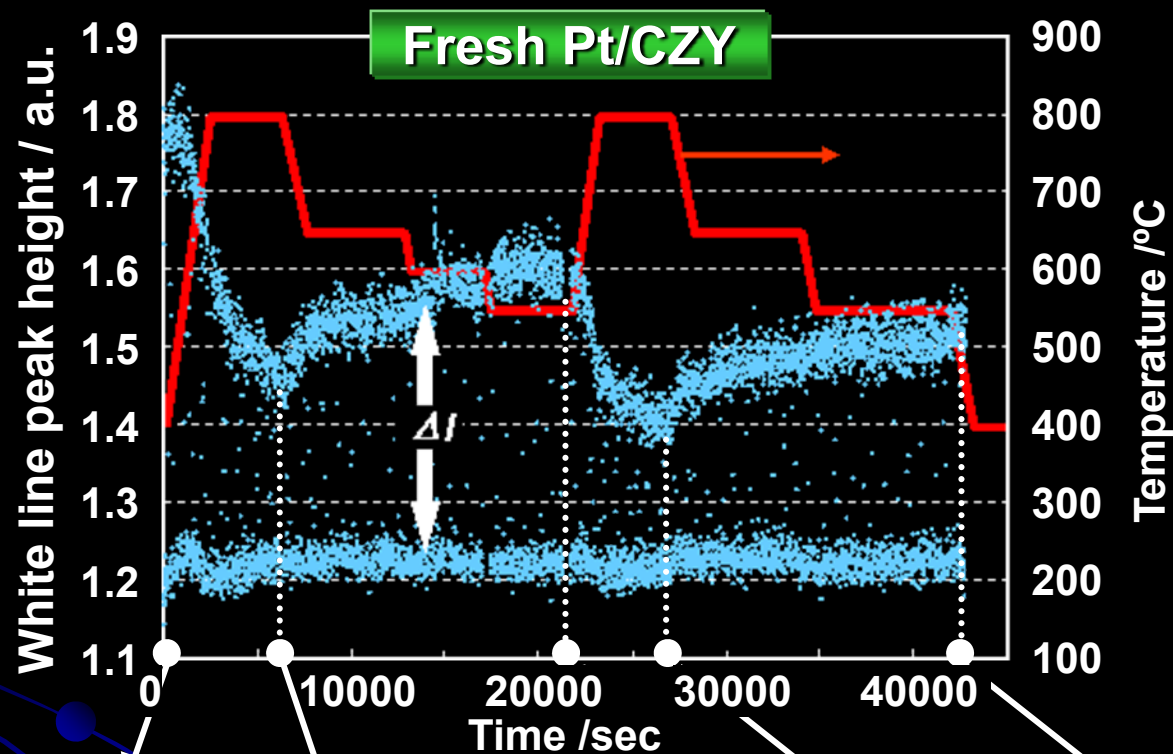
We observed only facile sintering in the Pt/Al₂O₃.



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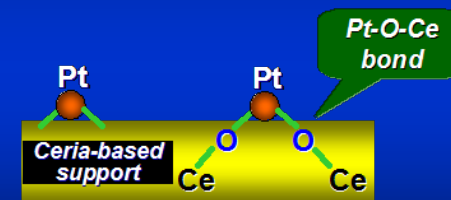
In-situ Dynamic Observation in Pt/CZY

3% H₂/He(60sec) ↔ 20 or 4 % O₂/He(60sec)



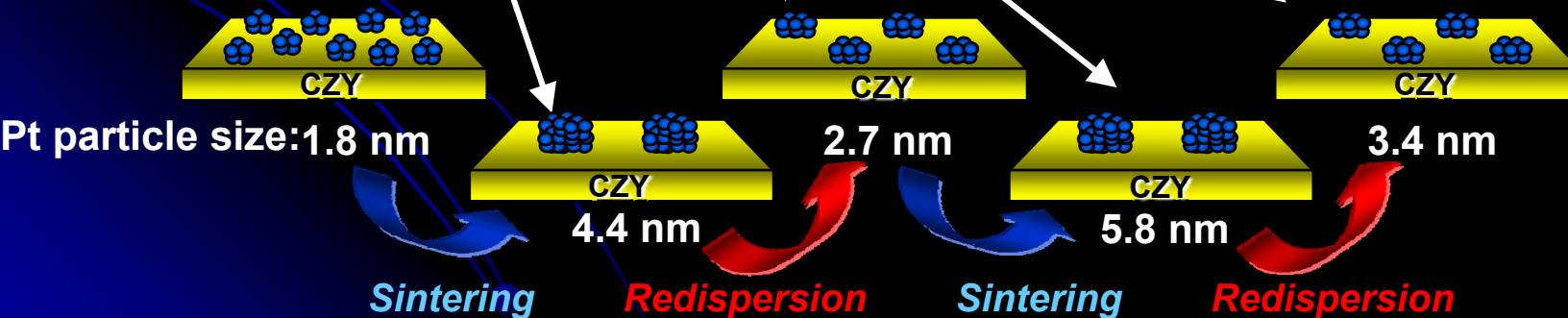
Strong Pt-oxide-ceria support interaction

Oxidative atmosphere



Y. Nagai et al.,
J. Catal. 242 (2006) 103

Redispersion experiments in hydrogen never resulted in increased dispersion.



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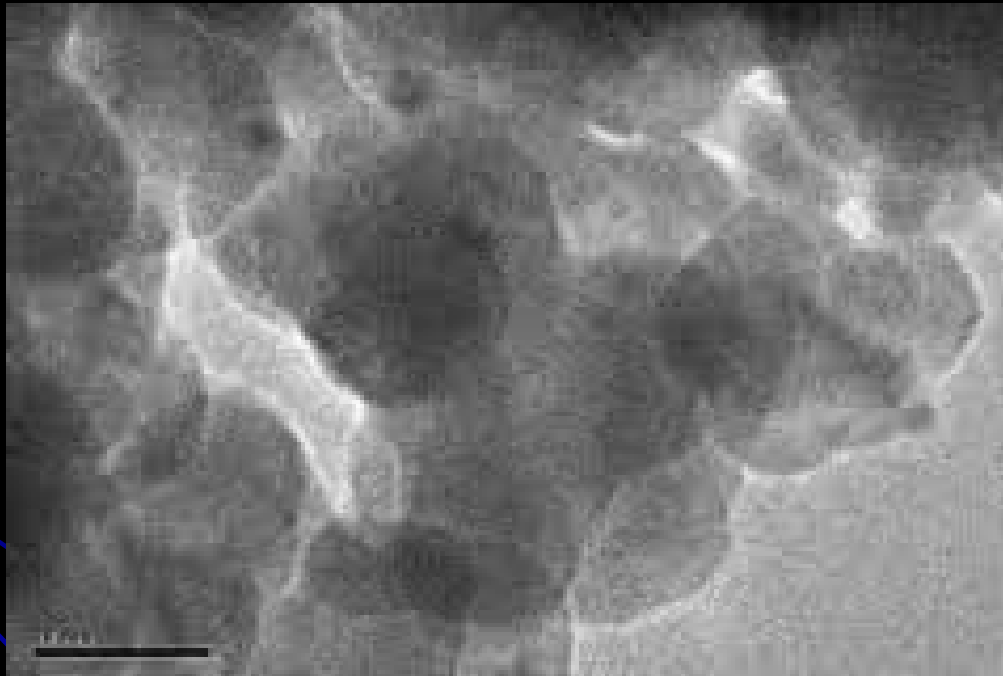
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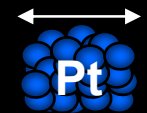


In-situ TEM Observation of Pt Redispersion

Pt redispersion process on CZY



6.7 nm



CZY

average starting
size 6.7nm

10 torr O₂ at 820°C



Oxidative redispersion

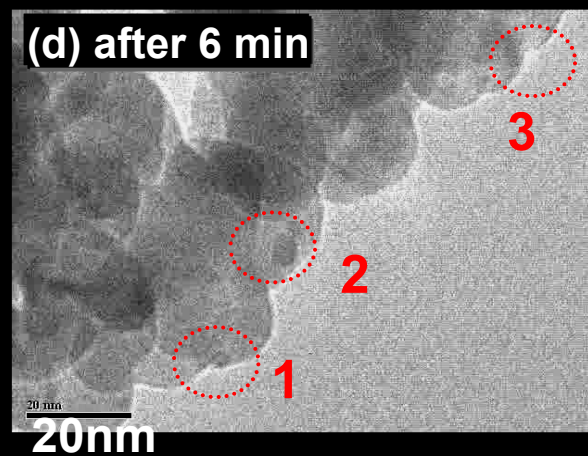
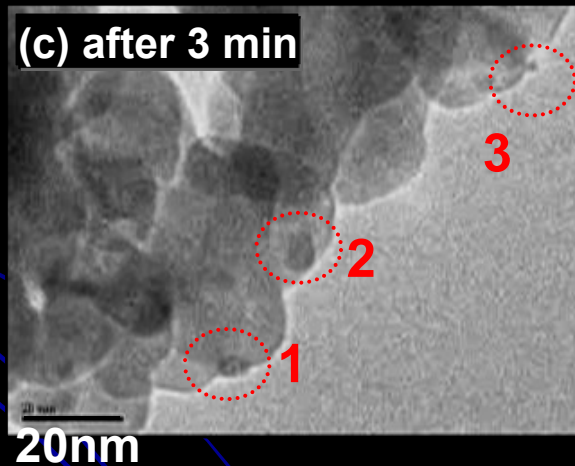
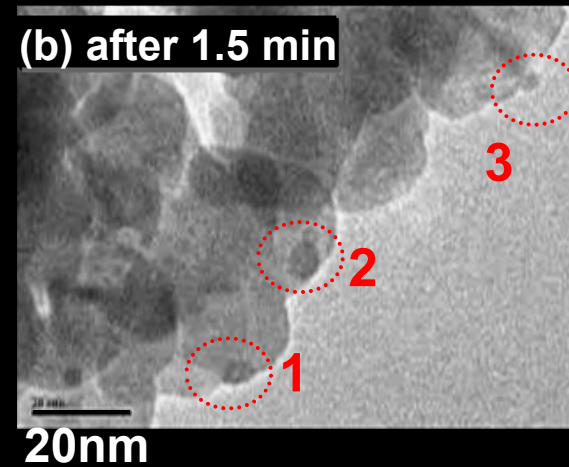
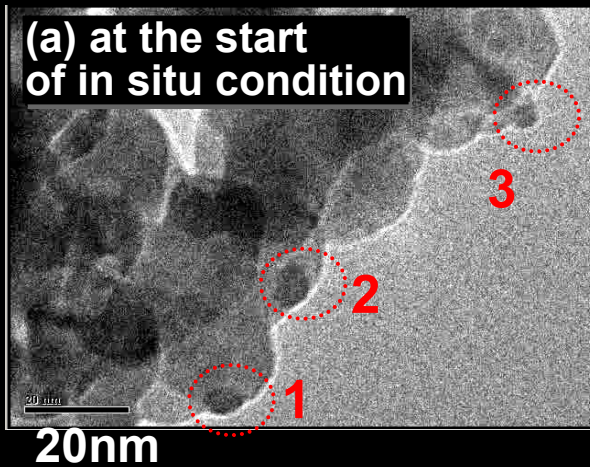
7 fold speed play
(7sec → 1sec)

*Pt sintering could not be observed in this field of view,
this movie captured the Pt redispersion process.*



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Snapshots of Pt Redispersion

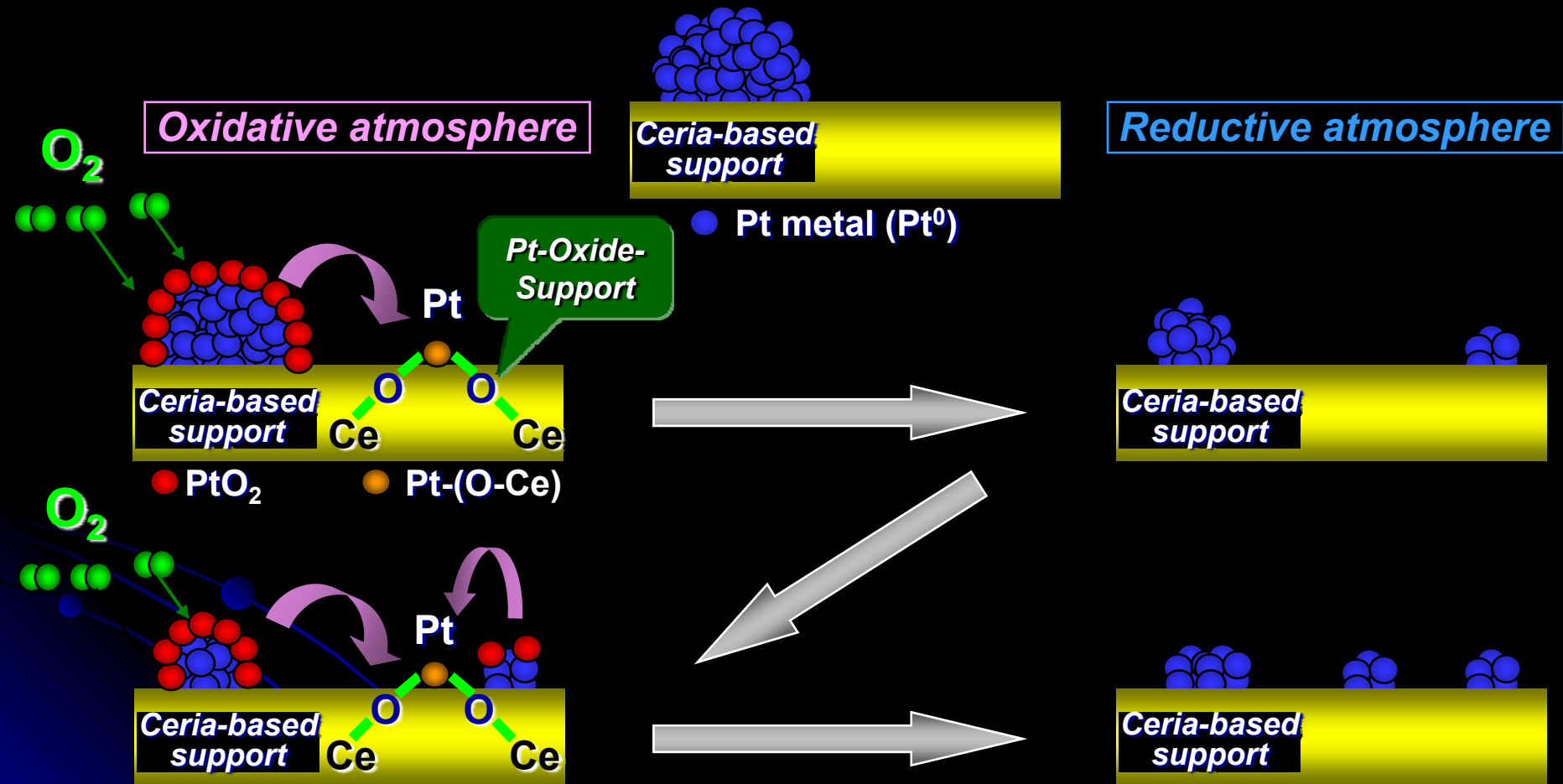


Pt particle size in the circle 1 and 2 decreased remarkably, and the Pt particle in the circle 3 disappeared. Crystallites migration and splitting was not observed.



Proposed mechanism of Pt redispersion

Sintered Catalyst



The Pt redispersion process proceeds by the repetition of the atomic migration of Pt oxide from the surface of large metal particles and the reduction of Pt-O-support species. (Not Pt crystallites migration)



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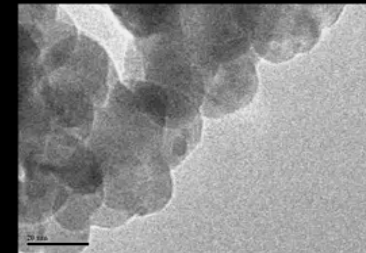
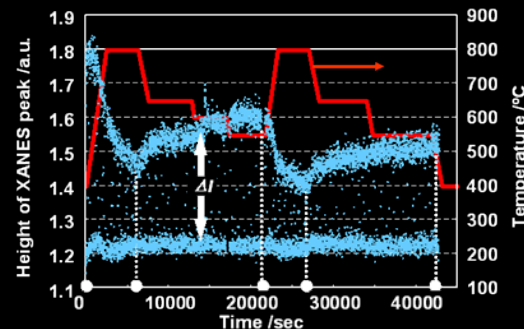
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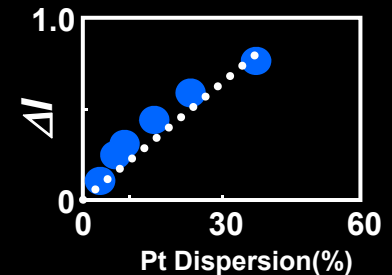
5. Future view

Significance

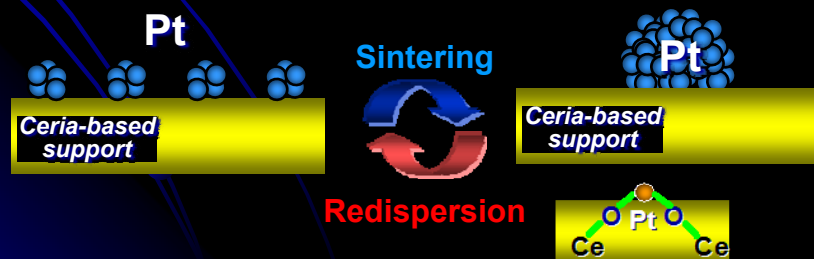
- a. Advanced technique for dynamic observation
in-situ time-resolved Turbo-XAS
in-situ TEM



- b. Novel method
in-situ dynamic measurement of Pt particle size
using Pt L₃-edge XANES

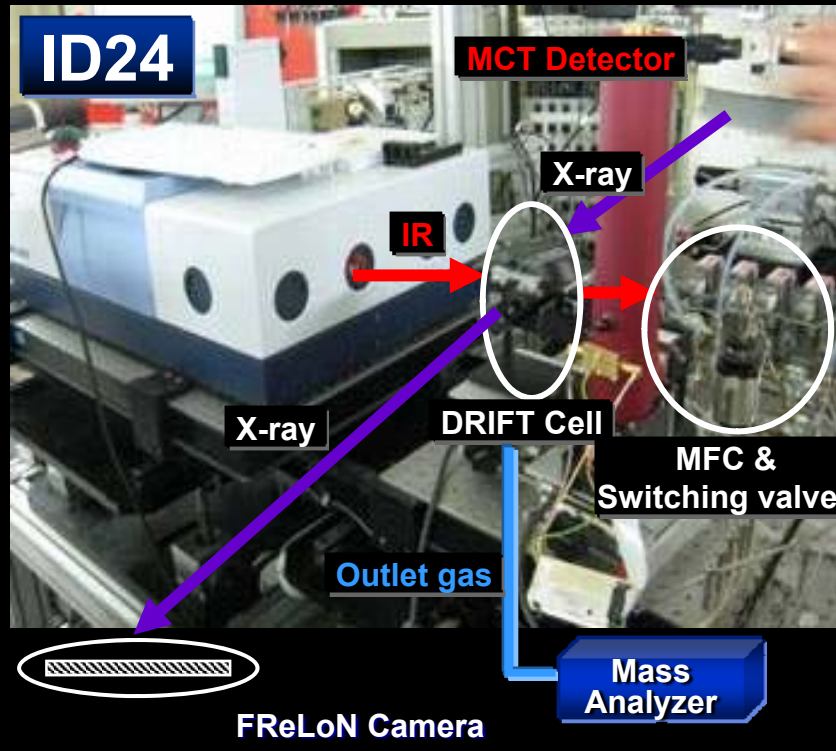
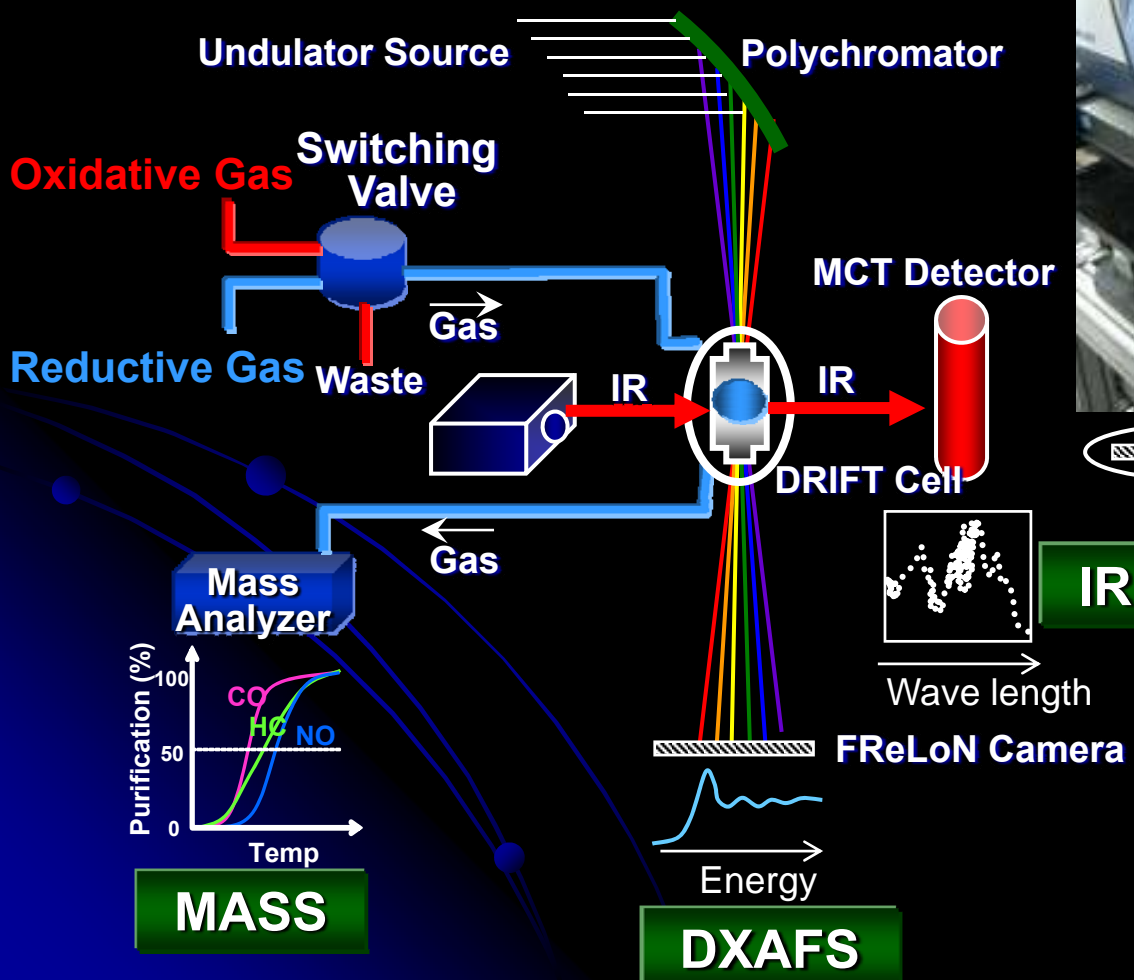


- c. New discovery
Reversible sintering/redispersion in simulated exhaust flowing
(effective Pt redispersion in the absence of Cl)
An atomic migration model accounts for the observed
redispersion through the strong Pt-oxide-support interaction.



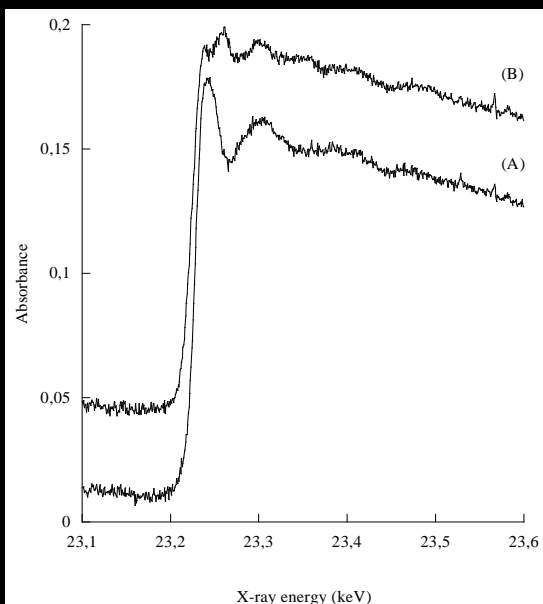
Future View

DXAFS - IR - Mass



Designed by M. A. Newton at ID24.

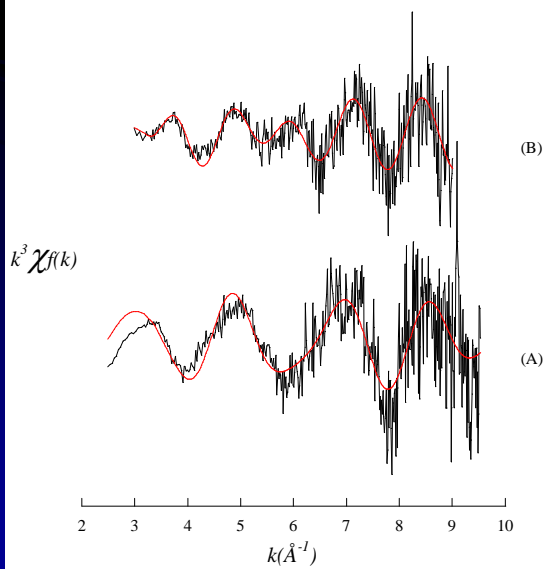
Now at 1wt% Rh.....



T = 573K

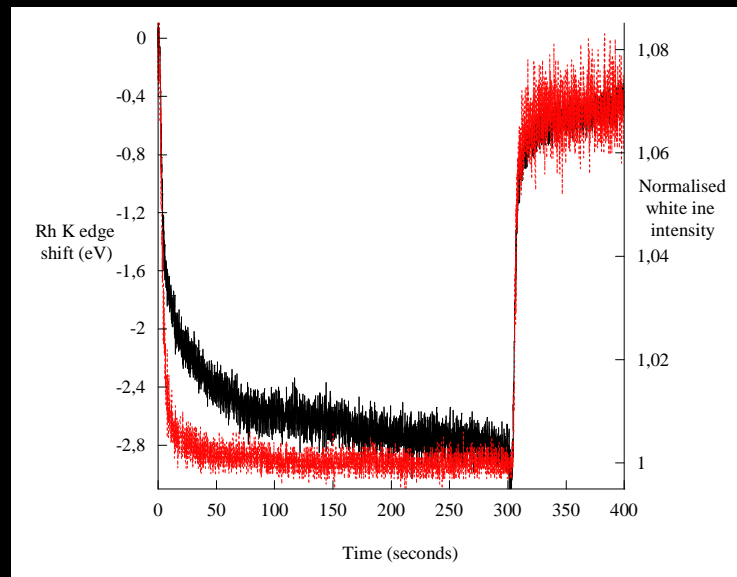
Reducing feed =
300ppm NO + 300ppm CO
78msecs/per spectrum

**A big improvement
on previous achievements
at Rh**

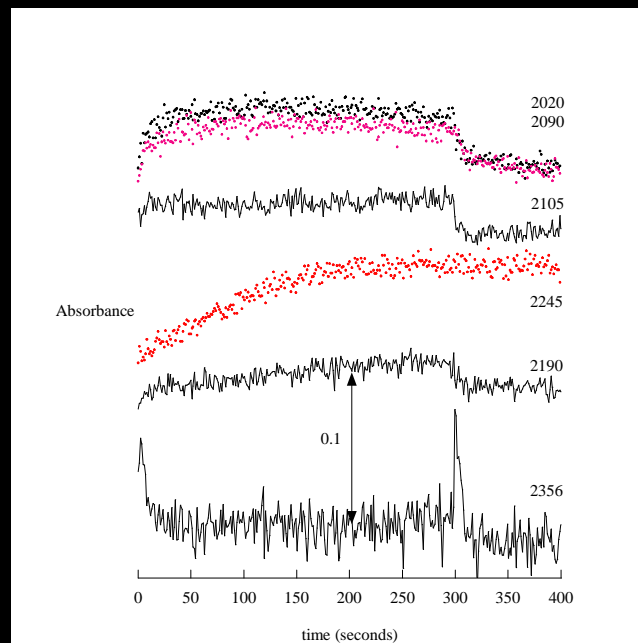


**A direct result of a
better
understanding of the
Tolerances w.r.t
instability
the beamline
optics have to work
within
for these systems**

Rh K edge position and white line intensity



Adsorbate behaviour (DRIFTS)



Obtainable Information

MASS

*Catalytic activity,
Reaction kinetics*

Reactant gas

IR

*Absorbed species and
Surface information on
active site*

DXAFS

*Dynamic Structural and
electronic changes of
catalytic active site*

Precious metal

Support

*Synchronised DXAFS-IR-Mass system is an “**Ultimate Operando System**” for heterogeneous catalysis.*

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Acknowledgments

ESRF

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TOYOTA Central R&D

Takamasa Nonaka, Satoshi Yamaguchi, Naoki Takahashi, Yoshiki Seno

~Towards a clean future~

