### Structure Refinements of II-VI Semiconductor Nanoparticles based on PDF Measurements

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#### Synthesis: Se in Trioctylphosphine + ZnEt<sub>2</sub> into Hexadecylamin at 310 C



Norris, D.J., Yao, N., Charnock, F.T. & Kennedy, T.A. (2001). *Nano Lett.* **1**, 3-6.

Rietveld Refinement: Zincblende Structure

a =4.00 Å Zn-Se = 2.45 Å FWHM<sub>111</sub>=3.3 Size ~ 26 Å  $R_{wp} = 14\%$ 

no fit at 311, high order hkl disordered material



# ZnSe Nanoparticles Fitting by Debye

Debye formula :

$$< |F(h)|^{2} > = \Sigma_{j} f_{j}^{2} + \Sigma_{i} \Sigma_{j,j \neq i} f_{i} f_{j} \sin(2\pi h r_{ij}) / (2\pi h r_{ij})$$

Sum over all atom pairs — no restrictions on sample structure

> open to finite particle with any shape defects like stacking faults etc.

# creating ZnSe Nanoparticles



create a large single Wurtzite layer A/B

Stack along c (with faults)Cut to proper sizeCalculate powder pattern

Repeat and average

Repeat with new set of parameter

using a Differential Evolutionary Scheme

 $<sup>\{110\}</sup>$  and  $\{001\}$ 



### **Powder Diffraction**

#### Powder diffraction pattern of a nanoparticle ZnSe



#### **Rietveld Refinement**

large background compared to Bragg reflections at higher  $2\Theta$ 

defects, organic ligands sample environment

Accurate Background estimation very difficult

limited information content, high correlation between parameters defects and size not well treated

very limited  $2\Theta$  range with significant reflections

small particle size defects

high uncertainties of structural parameters

#### PDF

#### Pair Distribution Function

essentially a Fourier transformation of the full powder diffraction pattern



Information about ordered and disordered structure



Information in direct space

old technique	B.E. Warren X-ray Diffraction (1969)	
does not require Bragg reflections		
old applications	glasses and liquids	
modern applications	<ul> <li>crystalline materials</li> <li>T. Egami and S.J.L. Billinge</li> <li>Underneath the Bragg Peaks</li> <li>Pergamonn (2003)</li> <li>S.J.L. Billinge and M.F. Thorpe (Eds.)</li> <li>Local Structure from Diffraction</li> <li>Plenum (1998)</li> </ul>	
requires modern syn neu	nrotron or laboratory: silver or tungsten sources	

#### **PDF Data Collection**

collect powder pattern to high  $2\Theta$  with high energy X-ray radiation

BW5 HASYLAB, DESY, Germany



www-hasylab.desy.de

http://lansce.lanl.gov/lujan/instruments/NPDF/index.html

#### ZnSe experimental PDF



Data treatment as in Korsounski et al., J. Appl. Cryst. 36, 1389 (2003)

Neder et al. phys. stat. sol. (c) 4, 3221 (2007)

#### ZnSe: Comparison to crystalline ZnSe



identical experimental conditions for both samples

### ZnSe experimental PDF



### ZnSe experimental PDF





#### ZnSe: Comparison to crystalline ZnSe



#### ZnSe: Comparison to crystalline ZnSe







#### structural coherence



loss of coherence due to stacking faults ~8 to 10 monolayers = 4 to 5 unit cells along c = 24 to 30 Å



#### Nanoparticle with core and stabilizing molecules



#### Algorithms for PDF Simulation of Nanoparticles

#### Simulate a crystal of N\*M\*O cells

calculated PDF with periodic boundary conditions multiply PDF by suitable shape function

Howell et al., Phys. Rev. B **73**, 094107 (2006) Kodama et al., Acta. Cryst. A **62**, 444 (2006)

#### Simulate a finite nanoparticle

calculate PDF from finite model correct shape of -4  $\pi \rho_0 r$  line

Neder et al. J. Phys.: Condens. Matter **17**, S125 (2005) Neder et al. phys. stat sol. (c),**4**, 3221 (2007)

#### PDF Simulation of Nanoparticles; envelope function



PDF of periodic ZnSe

q<sub>max</sub>, q<sub>alpha</sub>, etc. taken from fit to crystalline sample

as above, PDF multiplied by envelope function for a sphere



 $PDF_{nano} = PDF_{crystal} * f_{e}(r,d)$ 

 $f_{e}(r,d) = 1 - 3/2 r/d + \frac{1}{2} (r/d)^{3}$ 

defects can be treated

limited to basic shapes

treats two different effects! finite particle size change of average number density

Howell et al., Phys. Rev. B 73, 094107 (2006); Kodama et al., Acta. Cryst. A 62, 444 (2006)

#### PDF Simulation of Nanoparticles; finite particle

Simulation of a single finite sized ZnSe particle



PDF calculated without periodic boundary conditions

q<sub>max</sub>, q<sub>alpha</sub>, etc. taken from fit to crystalline sample

open to any shape here elliptical shape!

defects can be treated

defects in a single simulation are **NOT** a true represenation for whole sample

#### requires assembly average

assemly average may include: defect distribution size/shape distribution

average PDF of 20 individual particles with stacking fault

Neder et al. phys. stat. sol. (c) 4, 3221 (2007)

#### PDF Simulation of Nanoparticles; finite particle

Simulation of a single finite sized ZnSe particle



Neder et al. phys. stat. sol. (c) **4**, 3221 (2007)

#### Nanoparticle with core and stabilizing molecules



#### PDF Simulation of Nanoparticles; finite particle

Simulation of a single finite sized ZnSe particle



#### PDF Simulation of Nanoparticles; finite particle

#### Simulation of a single finite sized ZnSe particle



sphere:

 $-4 \pi \rho_0 r * f_e(r,d) = -4 \pi \rho_0 r * [1 - 3/2 r/d + \frac{1}{2} (r/d)^3]$ 

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 $\{110\}$  and  $\{001\}$ 



choose parent choose difference vector add to donor base to get donor cross-over between parent and donor compute cost function, keep better of parent/trial

Price, Storn & Lampinen Differential Evolution, Springer

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a 3.987 Å c 6.493 Å



ratio  $d_c/d_{ab} = 1.2$ elliptical shape

Stacking fault: 0.7

No distinction: prismatic vs spherical crystal

#### structural coherence



loss of coherence due to stacking faults ~8 to 10 monolayers = 4 to 5 unit cells along c = 24 to 30 Å



#### PDF Simulation of Nanoparticles; finite particle

#### Simulation of a single finite sized ZnSe particle



sphere:

 $-4 \pi \rho_0 r * f_e(r,d) = -4 \pi \rho_0 r * [1 - 3/2 r/d + \frac{1}{2} (r/d)^3] \qquad d = 1/3 (2*24 + 31) \text{ Å}$ 

#### CdSe Nanoparticles (Billinge)



a 4.303 Å c 6.997 Å non ideal tetrahedron z(Zn) = 0.382 ÅB iso = 2.3!size a-b=35(2) Å size c = 32(2) Å ratio  $d_c/d_{ab} = 0.9$ almost spherical shape Stacking fault: 0.28 density = 0.024first peak width = 0.56

scale = 0.85

 $\begin{array}{l} \delta \ = 0.00028 \\ \gamma \ = 0.08 \\ Q_{max} \ = 19 \ \text{\AA}^{\text{-1}} \end{array}$ 

### CdSe/ZnS Core/Shell particles



Core: CdSe ~3.2 nm Ø Shell: ZnS ~1 layer Stabilizer: TOPO

Band gap ZnS > CdSe efficient luminosity quantum confinement

#### Structure of Core / Shell ?

Epitaxial growth? 11% lattice mismatch!





Yu et al. Nano Lett., 5 (4), 565, 2005






anomalous powder diffraction ==> chemically selective structure info

#### CdSe/ZnS experimental PDF



#### CdSe/ZnS experimental PDF



#### CdSe/ZnS experimental PDF



narrow symmetrical first peaks

no indication of interaction



no significant differences CdSe core like crystalline structure







Elliptical CdSe core with stacking faults

a, c, z, B Ra, Rc, ρ

ZnS shell consisting of semi spherical subunits with stacking faults size distribution a, o Ra

a, c, z, B Ra, σR, ρ

Shell particles placed randomly at core surface, locally epitaxial **N** 









Carbon – Carbon distances

lattice constants as in bulk core and shell

high stacking fault probability core more wurtzite like 35% shell highly disorderd 50%

37 Å \* 39 Å radius core 10 Å thickness shell

no noticeable interaction between core and shell



Yu et al. Nano Lett., 5 (4), 565, 2005

#### stacking faults in II-VI nanoparticles

- ZnO Wurtzite 18%
- ZnSe Zincblende 30%

CdSe/ZnS core shell core Wurtzite 35% shell Zincblende 50%





# cubic closed packed stacking of tetrahedra

Zincblende

hexagonal close packed stacking of tetrahedra

Wurtzite



only minor differences in bond lengths



only minor differences in bond lengths additional different distances in Wurtzite all Zincblende distances also in Wurtzite!



only minor differences in bond lengths additional different distances in Wurtzite all Zincblende distances also in Wurtzite!





~8 to 10 monolayers = 4 to 5 unit cells along c = 24 to 30 Å

spherical II-VI nanoparticle 30 Å diameter ⇒ 660 atoms 25 Å diameter ⇒ 380 atoms layered structure layers identical in Wurtzite and Zincblende





# cubic closed packed stacking of tetrahedra

Zincblende

hexagonal close packed stacking of tetrahedra

Wurtzite

#### **Common building principles**





~ 40 atoms ~ 100 atoms

Common structural characteristics: layered structure layers identical in Wurtzite and Zincblende  $\Rightarrow$  high stacking fault probability 30% probability  $\Rightarrow$  3 to 4 faults

No relaxation of first neighbour distances Relaxation of second neighbour distance

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Debye formula













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= N c<sub>J</sub>  $\sum_{J} f_{J}^{2} + 2 \sum_{I} \sum_{J} f_{i} f_{j} \sum_{i} \sum_{j,j > i} \sin(2\pi h r_{ij}) / (2\pi h r_{ij}) / (2\pi h r_{ij})$ 

# ZnO Nanoparticles Fitting by Debye

```
< |F(h)|^{2} > = N c_{J} \Sigma_{J} f_{J}^{2} + 2 \Sigma_{I} \Sigma_{J} f_{i} f_{j} \Sigma_{i} \Sigma_{j,j>i} sin (2\pi h r_{ij}) / (2\pi h r_{ij})
```

```
for all atom i
for all atoms j > i
compile distance r_{ij} into histogram for type IJ
compile relative fraction of atoms type I
for all atom pairs IJ
for all h
multiply histogram by sin (2\pi h r_{ij}) / (2\pi h r_{ij}) (from lookup table)
multiply by 2*f_i f_j
for all atom type I
```

```
for all h
add f_i^2 * relative amount
```