



Synthesis process of inorganic nanoparticles

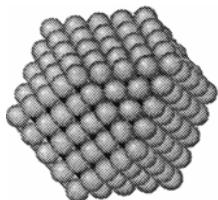
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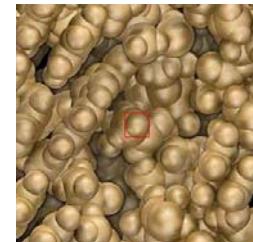
Universidad Nacional Autónoma de México.
Instituto de Investigaciones en Materiales.

Outline

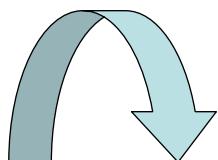
- Nanoparticles.
- Techniques of Synthesis.
- Ru-Sr-Gd-Cu-O system.
- Sr-Fe-(Mo,Re)-O system.
- CeO₂ compound.
- MNbO₃, M = Li, K, Na system.
- AConO_y, A = Li, Na, K, Ca and Sr and n = 1, 2, 4 system.



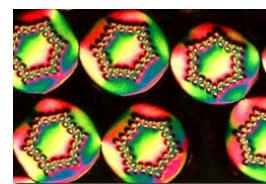
Nanoparticles



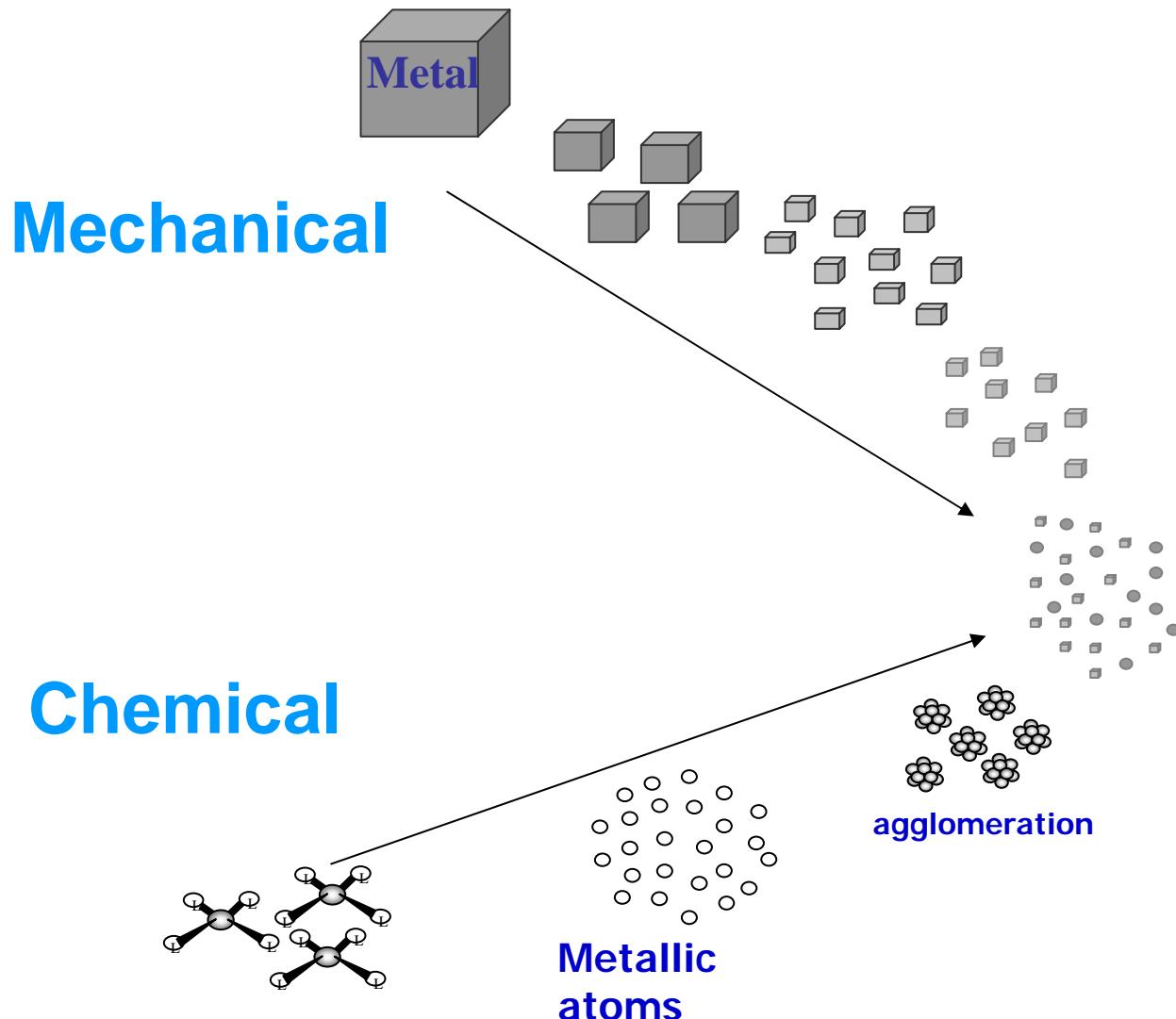
HIGH IMPACT:



- Superconductivity.
- Catalysis.
- Electric properties.
- Magnetic properties
- Thermoelectric properties.
- Optical properties.



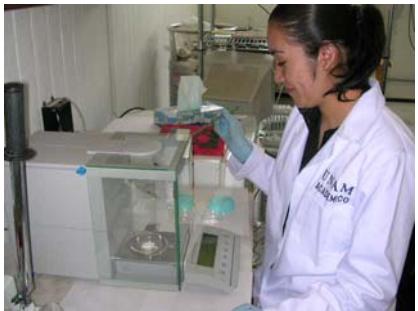
Synthesis techniques of nanoparticles.



Chemical synthesis:

- Sol-gel polymerized by acrylamide via microwave. 
- Hydrothermal synthesis.
- Decomposition of organometallic precursors.

Sol-gel polymerized by acrylamide via microwave.



pH



sol



gel



acrylamid

Continuation of the chemical process of Sol-gel polymerized by acrylamide via microwave.

gel



micro oven

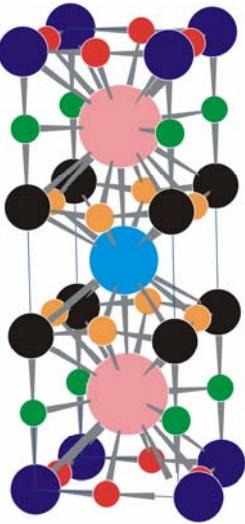


xerogel



Preliminary results.

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RuO
SrO
CuO
Gd
CuO
SrO
RuO

Ru-Sr-Gd-Cu-O system.

(Ru, Cu) -Sr-Gd-Cu-O.

Ru -(Sr,Ca)-Gd-Cu-O.

Ru -Sr-(Gd,Ln)-Cu-O: Ln = Dy, Ho, Er, Yb and Lu.

Prepared by sol-gel gelation with acrylamide via microwave.

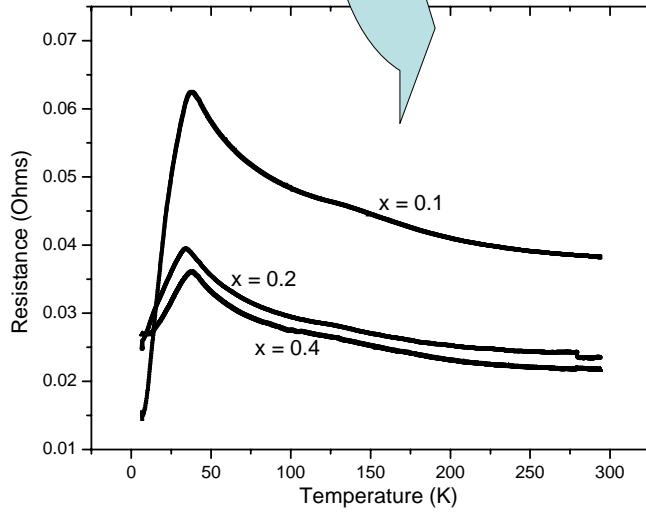


Fig. 5. R vs T for $Ru_{1-x}Sr_2GdCu_{2+x}O_8$ system

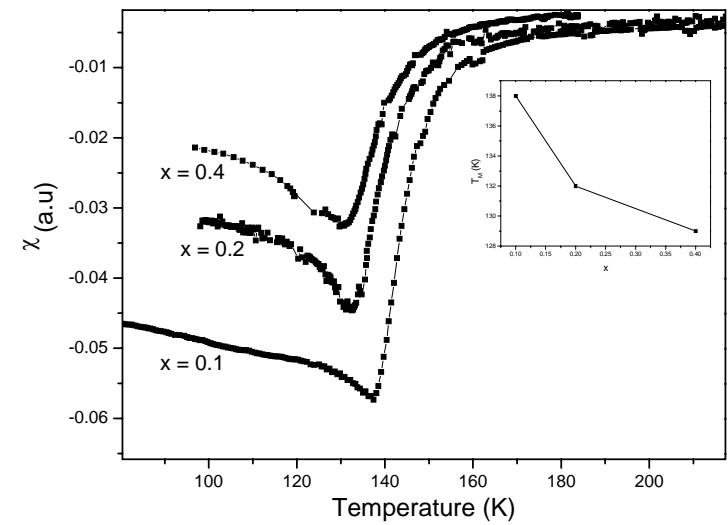
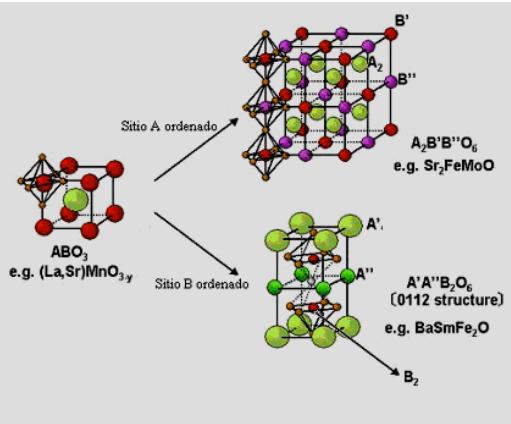


Fig. 7. χ_{ac} vs T of $Ru_{1-x}Sr_2GdCu_{2+x}O_8$

Sr-Fe-(Mo,Re)-O system.



Gel



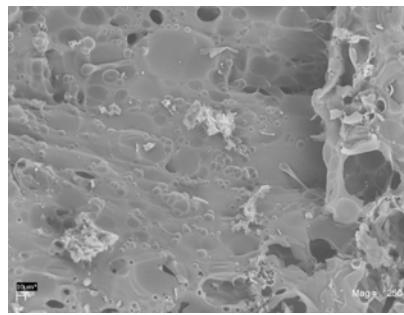
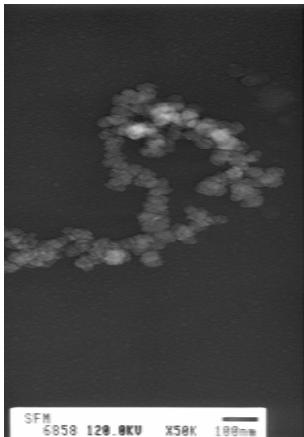
Sol



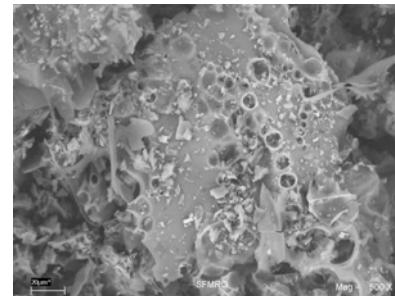
Microwave



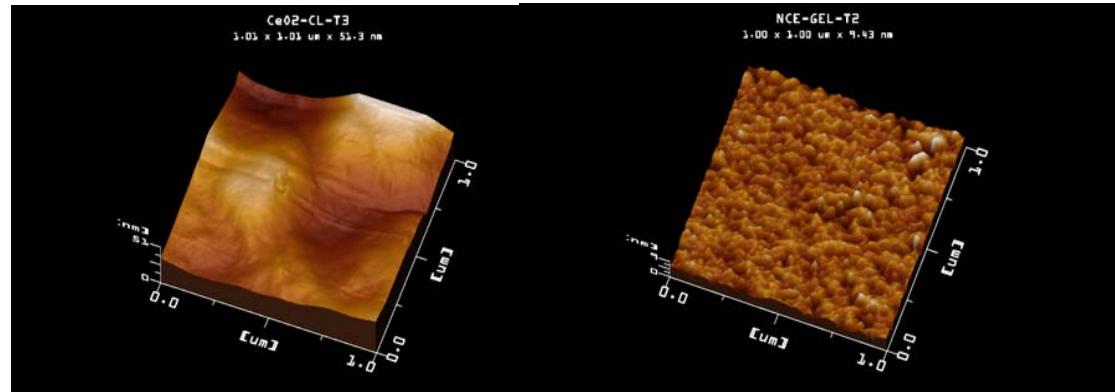
20-60 nm



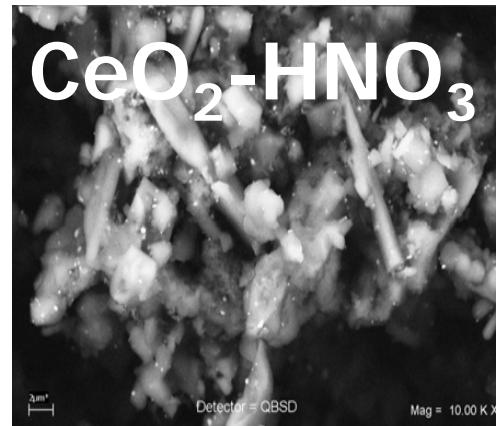
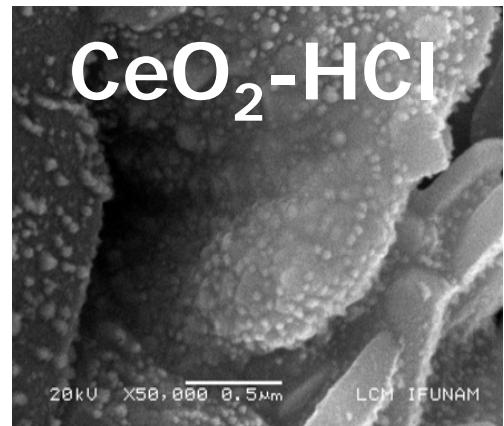
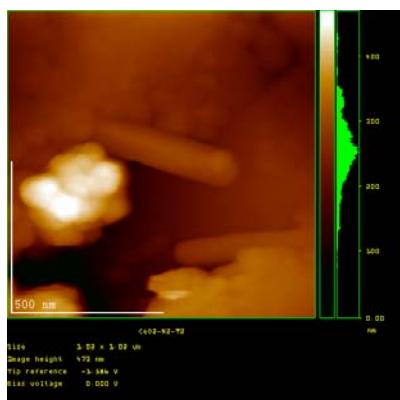
Xerogel



CeO₂ cubic compound.



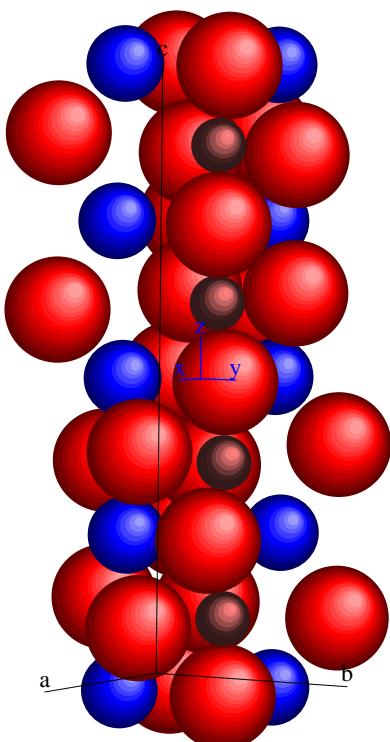
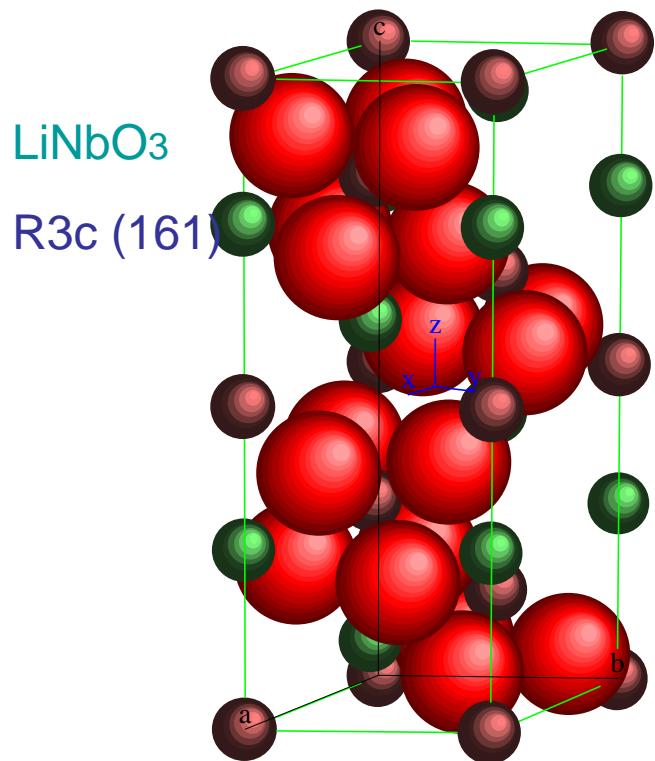
Prepared by sol-gel gelation with acrylate via microwave.



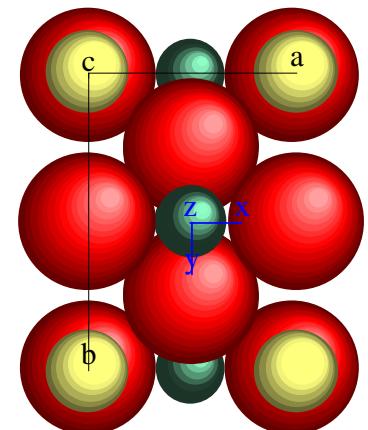
MNbO₃, M = Li, K, Na system.

Sol-gel polymerized by acrylamide via microwave.

Hydrothermal synthesis.



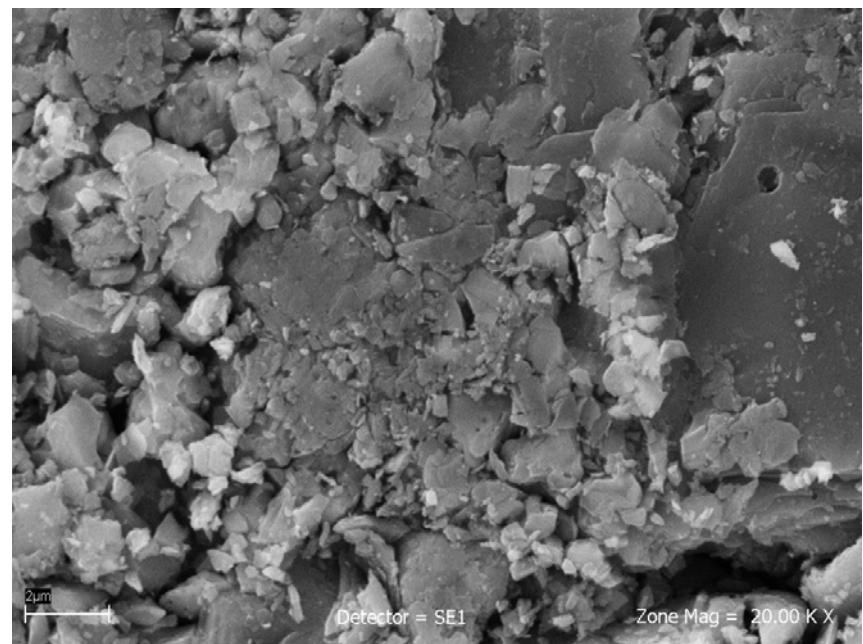
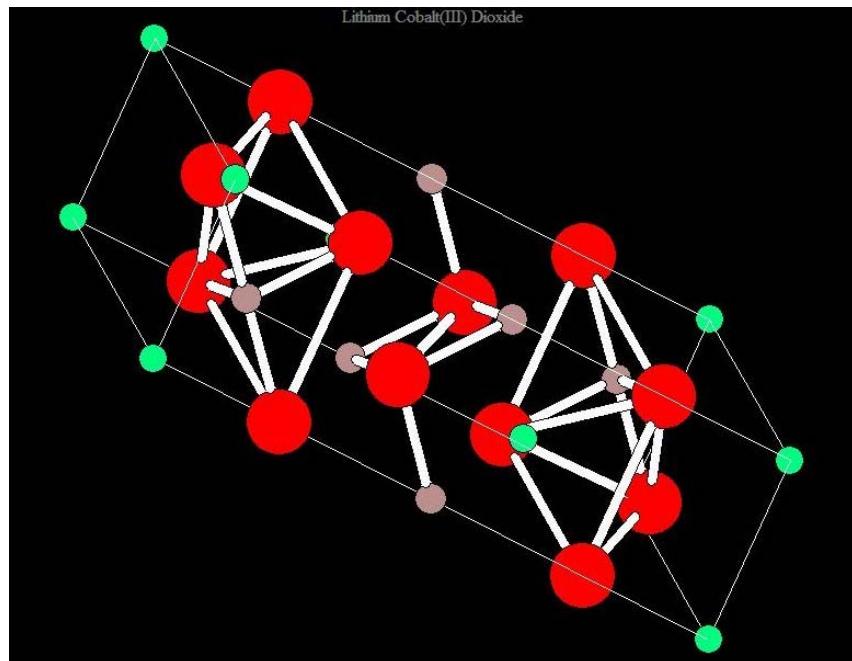
KNbO₃
Amm2 (38)



NaNbO₃
Pbcm (57)

AConO_y, A = Li, Na, K, Ca and Sr and n = 1, 2, 4 system.

LiCoO₂



Collaborators

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